## General Chemistry Unit 05: States of Matter

Author: Joanna Smithback

Published 2014

# Create, Share, and Discover Online Quizzes. 

QuirOwercom is an intuitive and powertal online quiz coeator lownmers


How to Analyze Stock
By Kosorlbohim

1monthogo
12 Repponse


Pre Employment Enzlich
BrKathoinojonniton
5montheago
19Reponse


Lean Startup Quiz
By Kosarlbuhim
2monthago
16Reponas
ollabrikion

## Powered by QuizOver.com

## The Leading Online Quiz \& Exam Creator

Create, Share and Discover Quizzes \& Exams
http://www.quizover.com

## Disclaimer

All services and content of QuizOver.com are provided under QuizOver.com terms of use on an "as is" basis, without warranty of any kind, either expressed or implied, including, without limitation, warranties that the provided services and content are free of defects, merchantable, fit for a particular purpose or non-infringing.

The entire risk as to the quality and performance of the provided services and content is with you.

In no event shall QuizOver.com be liable for any damages whatsoever arising out of or in connection with the use or performance of the services.

Should any provided services and content prove defective in any respect, you (not the initial developer, author or any other contributor) assume the cost of any necessary servicing, repair or correction.

This disclaimer of warranty constitutes an essential part of these "terms of use".

No use of any services and content of QuizOver.com is authorized hereunder except under this disclaimer.

The detailed and up to date "terms of use" of QuizOver.com can be found under:
http://www.QuizOver.com/public/termsOfUse.xhtml

Joanna Smithback, Ph.D. and Rachel Lerebours. General Chemistry I. The Saylor Foundation, http://www.saylor.org/courses/chem101/

Creative Commons License

Attribution-NonCommercial-NoDerivs 3.0 Unported (CC BY-NC-ND 3.0)
http://creativecommons.org/licenses/by-nc-nd/3.0/

You are free to:

Share: copy and redistribute the material in any medium or format

The licensor cannot revoke these freedoms as long as you follow the license terms.

Under the following terms:

Attribution: You must give appropriate credit, provide a link to the license, and indicate if changes were made. You may do so in any reasonable manner, but not in any way that suggests the licensor endorses you or your use.

NonCommercial: You may not use the material for commercial purposes.

NoDerivatives: If you remix, transform, or build upon the material, you may not distribute the modified material.

No additional restrictions: You may not apply legal terms or technological measures that legally restrict others from doing anything the license permits.

## Table of Contents

Quiz Permalink: http://www.quizover.com/question/unit-05-states-of-matter-by-joanna-smithback-saylor-foundat-general Author Profile: http://www.quizover.com/user/profile/joanna.smithback

1. Unit 05: States of Matter
(5) Powered by QuizOver.com - http://www.quizover.com

QuizOver.com is the leading online quiz \& exam creator
Copyright (c) 2009-2015 all rights reserved
4. Chapter: Unit 05: States of Matter

1. Unit 05: States of Matter Questions
(6) Powered by QuizOver.com - http://www.quizover.com

### 4.1.1. A sample of a certain gas has a volume of 222 mL at 695 mm Hg and $0 \ldots$

Author: Joanna Smithback
A sample of a certain gas has a volume of 222 mL at 695 mm Hg and $0^{\circ} \mathrm{C}$. What would be the volume of this same sample of gas if it were measured at 333 mm Hg and $0^{\circ} \mathrm{C}$ ?

Please choose only one answer:

- 894 mL
- 657 mL
- 463 mL
- 359 mL

Check the answer of this question online at QuizOver.com:
Question: A sample of a certain gas has a volume of Joanna Smithback @Saylor
Flashcards:
http://www.quizover.com/flashcards/a-sample-of-a-certain-gas-has-a-volume-of-joanna-smithback-saylor?pdf=3044
Interactive Question:
http://www.quizover.com/question/a-sample-of-a-certain-gas-has-a-volume-of-joanna-smithback-saylor?pdf=3044
4.1.2. For a given sample of gas molecules, the average kinetic energy dep...

Author: Joanna Smithback
For a given sample of gas molecules, the average kinetic energy depends only on the value of the
$\qquad$ -.

Please choose only one answer:

- pressure
- temperature
- volume
- moles

Check the answer of this question online at QuizOver.com:
Question: For a given sample of gas molecules the Joanna Smithback @Saylor
Flashcards:
http://www.quizover.com/flashcards/for-a-given-sample-of-gas-molecules-the-joanna-smithback-saylor?pdf=3044
Interactive Question:
http://www.quizover.com/question/for-a-given-sample-of-gas-molecules-the-joanna-smithback-saylor?pdf=3044
(8) Powered by QuizOver.com - http://www.quizover.com

QuizOver.com is the leading online quiz \& exam creator
Copyright (c) 2009-2015 all rights reserved
$\qquad$ .

## Author: Joanna Smithback

The phase change from solid to gas is called $\qquad$ .

Please choose only one answer:

- boiling
- efflorescence
- evaporation
- sublimation

Check the answer of this question online at QuizOver.com:
Question: The phase change from solid to gas is Joanna Smithback @Saylor General
Flashcards:
http://www.quizover.com/flashcards/the-phase-change-from-solid-to-gas-is-joanna-smithback-saylor-general?pdf=3044
Interactive Question:
http://www.quizover.com/question/the-phase-change-from-solid-to-gas-is-joanna-smithback-saylor-general?pdf=3044
(9) Powered by QuizOver.com - http://www.quizover.com

QuizOver.com is the leading online quiz \& exam creator
Copyright (c) 2009-2015 all rights reserved

### 4.1.4. If the temperature and pressure are kept constant during the proces...

Author: Joanna Smithback
If the temperature and pressure are kept constant during the process, how many liters of TiCl[sub]4[/sub] gas will be produced when 20.0 L of chlorine react with titanium, according to the reaction: Ti(s) $+2 \mathrm{Cl}[\mathrm{sub}] 2[/ \mathrm{sub}](\mathrm{g})$ TiCl[sub]4[/sub](g)?

Please choose only one answer:

- 5.00 L
- 10.0 L
- 20.0 L
- 40.0 L

Check the answer of this question online at QuizOver.com:
Question: If the temperature and pressure are kept Joanna Smithback @Saylor
Flashcards:
http://www.quizover.com/flashcards/if-the-temperature-and-pressure-are-kept-joanna-smithback-saylor?pdf=3044
Interactive Question:
http://www.quizover.com/question/if-the-temperature-and-pressure-are-kept-joanna-smithback-saylor?pdf=3044
4.1.5. In a phase diagram, if a sample is at low pressure and high tempera...

Author: Joanna Smithback
In a phase diagram, if a sample is at low pressure and high temperature, it is likely to be in what phase?

Please choose only one answer:

- Liquid
- Gas
- Solid
- Both A and C

Check the answer of this question online at QuizOver.com:
Question: In a phase diagram if a sample is at low Joanna Smithback @Saylor
Flashcards:
http://www.quizover.com/flashcards/in-a-phase-diagram-if-a-sample-is-at-low-joanna-smithback-saylor?pdf=3044
Interactive Question:
http://www.quizover.com/question/in-a-phase-diagram-if-a-sample-is-at-low-joanna-smithback-saylor?pdf=3044

### 4.1.6. Ionic crystals have the following properties.

Author: Joanna Smithback
Ionic crystals have the following properties.

Please choose only one answer:

- Long range order
- Amorphous
- Made up of neutral molecules
- None of the above

Check the answer of this question online at QuizOver.com:
Question: lonic crystals have the following properties Joanna @Saylor Foundat
Flashcards:
http://www.quizover.com/flashcards/ionic-crystals-have-the-following-properties-joanna-saylor-foundat?pdf=3044
Interactive Question:
http://www.quizover.com/question/ionic-crystals-have-the-following-properties-joanna-saylor-foundat?pdf=3044
(12) Powered by QuizOver.com - http://www.quizover.com

QuizOver.com is the leading online quiz \& exam creator
Copyright (c) 2009-2015 all rights reserved

### 4.1.7. Nitrogen gas has a pressure of 452 mm Hg . What is the pressure in a...

Author: Joanna Smithback
Nitrogen gas has a pressure of 452 mm Hg . What is the pressure in atmospheres?

Please choose only one answer:

- 4.52 atm
- 0.595 atm
- 0.452 atm
- 1.68 atm

Check the answer of this question online at QuizOver.com:
Question: Nitrogen gas has a pressure of 452 mm Hg Joanna Smithback @Saylor
Flashcards:
http://www.quizover.com/flashcards/nitrogen-gas-has-a-pressure-of-452-mm-hg-joanna-smithback-saylor?pdf=3044
Interactive Question:
http://www.quizover.com/question/nitrogen-gas-has-a-pressure-of-452-mm-hg-joanna-smithback-saylor?pdf=3044
(13) Powered by QuizOver.com - http://www.quizover.com

QuizOver.com is the leading online quiz \& exam creator
Copyright (c) 2009-2015 all rights reserved

### 4.1.8. The atmospheric pressure on a nice day is 751 mm Hg . What is this p...

Author: Joanna Smithback
The atmospheric pressure on a nice day is 751 mm Hg. What is this pressure in atmospheres?

Please choose only one answer:

- $\quad 0.751 \mathrm{~atm}$
- 7.51 atm
- 0.988 atm
- 1.01 atm

Check the answer of this question online at QuizOver.com:
Question: The atmospheric pressure on a nice day is Joanna Smithback @Saylor
Flashcards:
http://www.quizover.com/flashcards/the-atmospheric-pressure-on-a-nice-day-is-joanna-smithback-saylor?pdf=3044
Interactive Question:
http://www.quizover.com/question/the-atmospheric-pressure-on-a-nice-day-is-joanna-smithback-saylor?pdf=3044

### 4.1.9. The liquid to gas phase transition for the drying rainwater on a ro...

Author: Joanna Smithback
The liquid to gas phase transition for the drying rainwater on a road, following a rainstorm, is described to be what?

Please choose only one answer:

- Boiling
- Evaporation
- Sublimation
- None of the above

Check the answer of this question online at QuizOver.com:
Question: The liquid to gas phase transition for Joanna Smithback @Saylor General
Flashcards:
http://www.quizover.com/flashcards/the-liquid-to-gas-phase-transition-for-joanna-smithback-saylor-general?pdf=3044
Interactive Question:
http://www.quizover.com/question/the-liquid-to-gas-phase-transition-for-joanna-smithback-saylor-general?pdf=3044
4.1.10. The volume of a certain gas sample is 1150 mL at a temperature of $2 . .$.

Author: Joanna Smithback
The volume of a certain gas sample is 1150 mL at a temperature of $25^{\circ} \mathrm{C}$. At what temperature would that same gas sample have a volume of 1.53 L at constant pressure and mass?

Please choose only one answer:

- $-49.1^{\circ} \mathrm{C}$
- $124^{\circ} \mathrm{C}$
- $248^{\circ} \mathrm{C}$
- $397^{\circ} \mathrm{C}$

Check the answer of this question online at QuizOver.com:
Question: The volume of a certain gas sample is 11 Joanna Smithback @Saylor
Flashcards:
http://www.quizover.com/flashcards/the-volume-of-a-certain-gas-sample-is-11-joanna-smithback-saylor?pdf=3044
Interactive Question:
http://www.quizover.com/question/the-volume-of-a-certain-gas-sample-is-11-joanna-smithback-saylor?pdf=3044

### 4.1.11. The volume of a certain gas sample is 235 mL at a temperature of $25 \ldots$...

Author: Joanna Smithback
The volume of a certain gas sample is 235 mL at a temperature of $25^{\circ} \mathrm{C}$. At what temperature would that same gas sample have a volume of 310 mL at constant pressure and mass?

Please choose only one answer:

- $-47.0^{\circ} \mathrm{C}$
- $33.1^{\circ} \mathrm{C}$
- $69.4^{\circ} \mathrm{C}$
- $120^{\circ} \mathrm{C}$

Check the answer of this question online at QuizOver.com:
Question: The volume of a certain gas sample is 23 Joanna Smithback @Saylor
Flashcards:
http://www.quizover.com/flashcards/the-volume-of-a-certain-gas-sample-is-23-joanna-smithback-saylor?pdf=3044
Interactive Question:
http://www.quizover.com/question/the-volume-of-a-certain-gas-sample-is-23-joanna-smithback-saylor?pdf=3044
4.1.12. What is the pressure in atmospheres of a gas mixture that consists ...

Author: Joanna Smithback
What is the pressure in atmospheres of a gas mixture that consists of 0.200 moles of nitrogen and 0.300 moles of oxygen in a 1250 mL container at $0^{\circ} \mathrm{C}$ ?

Please choose only one answer:

- 0.00897 atm
- 0.897 atm
- $\quad 1.79 \mathrm{~atm}$
- 8.97 atm

Check the answer of this question online at QuizOver.com:
Question: What is the pressure in atmospheres of a Joanna Smithback @Saylor
Flashcards:
http://www.quizover.com/flashcards/what-is-the-pressure-in-atmospheres-of-a-joanna-smithback-saylor?pdf=3044
Interactive Question:
http://www.quizover.com/question/what-is-the-pressure-in-atmospheres-of-a-joanna-smithback-saylor?pdf=3044
4.1.13. What is the pressure in atmospheres of a gas mixture that consists ...

Author: Joanna Smithback
What is the pressure in atmospheres of a gas mixture that consists of 8.80 grams of nitrogen and 8.80 grams of carbon dioxide in a 2.01 liter container at $27^{\circ} \mathrm{C}$ ?

Please choose only one answer:

- 6.30 atm
- 3.85 atm
- 2.45 atm
- 0.971 atm

Check the answer of this question online at QuizOver.com:
Question: What is the pressure in atmospheres of a Joanna Smithback @Saylor
Flashcards:
http://www.quizover.com/flashcards/what-is-the-pressure-in-atmospheres-of-a-joanna-smithback-sayl-9423521?pdf=3044
Interactive Question:
http://www.quizover.com/question/what-is-the-pressure-in-atmospheres-of-a-joanna-smithback-sayl-9423521?pdf=3044
4.1.14. You have $O[s u b] 2[/ s u b]$ gas with a pressure of 0.32 atm . What is the...

Author: Joanna Smithback
You have $\mathrm{O}[s u b] 2[/$ sub] gas with a pressure of 0.32 atm . What is the gas pressure in mm Hg ?

Please choose only one answer:

- 240 mm Hg
- 0.24 mm Hg
- 0.0041 mm Hg
- 24 mm Hg

Check the answer of this question online at QuizOver.com:
Question: You have O sub 2 /sub gas with a pressure Joanna Smithback @Saylor
Flashcards:
http://www.quizover.com/flashcards/you-have-o-sub-2-sub-gas-with-a-pressure-joanna-smithback-saylor?pdf=3044
Interactive Question:
http://www.quizover.com/question/you-have-o-sub-2-sub-gas-with-a-pressure-joanna-smithback-saylor?pdf=3044

